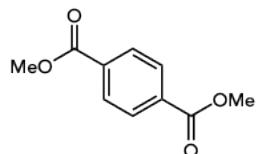


3

¹H NMR (500 MHz, CDCl₃)



$$\text{mg}(3) = 114.0 \text{ mg}$$

$$\text{MW}(3) = 274.36 \text{ g/mol}$$

$$\text{mg(std)} = 19.9 \text{ mg}$$

$$\text{MW(std)} = 194.19 \text{ g/mol}$$

$$\text{Molar ratio} = [12.04/3H]$$

$$\text{wt\%} = \frac{19.9 * 274.36 * \left(\frac{3.01}{3} * 4\right)}{114.0 * 194.19}$$

$$\text{wt\%} = 99.0\% \text{ pure}$$

